

## Production, oxidation, emission and consumption of methane by soils: A review

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**Abstract** – Methane emission by soils results from antagonistic but correlated microbial activities. Methane is produced in the anaerobic zones of submerged soils by methanogens and is oxidised into CO<sub>2</sub> by methanotrophs in the aerobic zones of wetland soils and in upland soils. Methanogens and methanotrophs are ubiquitous in soils where they remain viable under unfavourable conditions. Methane transfer from the soil to the atmosphere occurs mostly through the aerenchyma of aquatic plants, but also by diffusion and as bubbles escaping from wetland soils. Methane sources are mainly wetlands. However 60 to more than 90 % of CH<sub>4</sub> produced in the anaerobic zones of wetlands is reoxidised in their aerobic zones (rhizosphere and oxidised soil-water interface). Methane consumption occurs in most soils and exhibits a broad range of values. Highest consumption rates or potentials are observed in soils where methanogenesis is or has been effective and where CH<sub>4</sub> concentration is or has been much higher than in the atmosphere (ricefields, swamps, landfills, etc.). Aerobic soils consume atmospheric CH<sub>4</sub> but their activities are very low and the micro-organisms involved are largely unknown. Methane emissions by cultivated or natural wetlands are expressed in mg CH<sub>4</sub>·m<sup>-2</sup>·h<sup>-1</sup> with a median lower than 10 mg CH<sub>4</sub>·m<sup>-2</sup>·h<sup>-1</sup>. Methanotrophy in wetlands is most often expressed with the same unit. Methane oxidation by aerobic upland soils is rarely higher than 0.1 mg CH<sub>4</sub>·m<sup>-2</sup>·h<sup>-1</sup>. Forest soils are the most active, followed by grasslands and cultivated soils. Factors that favour CH<sub>4</sub> emission from cultivated wetlands are mostly submersion and organic matter addition. Intermittent drainage and utilisation of the sulphate forms of N-fertilisers reduce CH<sub>4</sub> emission. Methane oxidation potential of upland soils is reduced by cultivation, especially by ammonium N-fertiliser application. © 2001 Éditions scientifiques et médicales Elsevier SAS

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### 1. INTRODUCTION

Methane is the main hydrocarbon present in the atmosphere, with an average concentration of 1.7 ppm. Variations between the northern and southern hemispheres average 0.14 ppm and exhibit seasonal variations of about 0.03 ppm [55].

Despite a short residence time in the atmosphere (about 10 years), the CH<sub>4</sub> ability to absorb infrared radiation makes it 20 to 30 times more efficient than CO<sub>2</sub> as a greenhouse gas [17, 169]. Methane is chemically very reactive and is therefore involved in changes in the chemical composition of the atmosphere [34]. In particular, it reacts with hydroxyl

radicals in the troposphere, reducing its oxidative power and ability to eliminate pollutants such as chloro-fluoro carbons (CFCs), and leading to the production of other greenhouse gases (ozone, CO, CO<sub>2</sub>). In the stratosphere, such reactions produce water vapour, which is involved in the destruction of the stratospheric ozone layer, the natural barrier against detrimental solar radiations. Methane is considered the second or the third greenhouse gas after CO<sub>2</sub> and CFCs [121, 138].

Annual CH<sub>4</sub> emission, estimated from the analysis of air trapped in polar ice, were 180 Tg·year<sup>-1</sup> during the 15th century (1 Tg = 10<sup>12</sup> g) and 200 Tg·year<sup>-1</sup> at the beginning of the 18th century [102]. The recent estimates of the International Panel for Climate Changes (IPCC) [88] are around 300 Tg in 2000, and between 400 and 600 Tg in 2010.

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Atmospheric CH<sub>4</sub> is mainly (70–80 %) of biological origin. It is produced in anoxic environments, including submerged soils, by methanogenic bacteria during the anaerobic digestion of organic matter. Methane is mainly eliminated in the troposphere through oxidation by OH• radicals, according to the reaction: CH<sub>4</sub> + OH• → CH<sub>3</sub>• + H<sub>2</sub>O. In the stratosphere, CH<sub>4</sub> also reacts with chlorine (originating from CFCs) according to the reaction: CH<sub>4</sub> + Cl• → HCl + CH<sub>3</sub>•. Methane is also eliminated in soils by microbial oxidation, which takes place in the aerobic zone of methanogenic soils (methanotrophy) and in upland soils, which oxidise atmospheric methane. Soils most efficient in methanotrophy are generally those from sites that are often submerged or water-saturated and where a significant methanogenic activity develops at intervals [146]. Ricefield soils, peat soils [204] and soils from landfills [238] often exhibit very high potential methanotrophic activities but in such environments, where anaerobiosis predominate, the balance between CH<sub>4</sub> production and oxidation is usually positive.

An environment is a CH<sub>4</sub> source when the balance between production by methanogenic bacteria and consumption by methanotrophic bacteria is positive, leading to CH<sub>4</sub> emission. When the balance is negative, the environment is a CH<sub>4</sub> sink.

Natural CH<sub>4</sub> sources are considered responsible for about 30 % of total emissions. Wetland soils (swamps, bogs, etc.) are the main natural source with an estimated emission of 100–200 Tg·year<sup>-1</sup>. Other sources are oceans, some forest soils, termites and wild ruminants (*table I*). About 70 % of CH<sub>4</sub> emissions are of human origin. Domesticated ruminants (65–100 Tg·year<sup>-1</sup>) and ricefields (25–150 Tg·year<sup>-1</sup>) are responsible for 15–40 % of total emissions, therefore agriculture is the main anthropic source of CH<sub>4</sub>.

Because of their economical importance and high potential as CH<sub>4</sub> source, ricefields have been the most studied methanogenic ecosystems. They are also the most suitable model to study CH<sub>4</sub> emission because methanogenesis and methanotrophy are very active

and all modes of CH<sub>4</sub> transfer occur in ricefields. Assuming an annual emission of 50 Tg CH<sub>4</sub> by ricefields, the production of 1 kg rice corresponds to the emission of 100 g CH<sub>4</sub>.

As the sources of atmospheric CH<sub>4</sub> are closely related to human activities, it is theoretically possible to control them. According to Thompson et al. [208], the global temperature increase could be reduced by 25 % if CH<sub>4</sub> emissions could be stabilised.

Temperate and tropical oxic soils that are continuously emerged and exposed to atmospheric concentrations of CH<sub>4</sub> are CH<sub>4</sub> sinks. They usually exhibit low levels of atmospheric CH<sub>4</sub> oxidation but, because of the large areas they cover, they are estimated to consume about 10 % of the atmospheric CH<sub>4</sub> (*table I*). Among upland soils, forest soils are probably the most efficient CH<sub>4</sub> sink. Atmospheric CH<sub>4</sub> oxidation also occurs in extreme environments such as deserts and glaciers, in the floodwater of submerged soils and in river waters.

According to IPCC estimates [87], natural and cultivated submerged soils (landfills not included) contribute about 55 % of the CH<sub>4</sub> emitted into the atmosphere, corresponding to 175 Tg·year<sup>-1</sup>, while upland soils are responsible for 6 % of the CH<sub>4</sub> consumption, corresponding to 30 Tg·year<sup>-1</sup> (*table I*). Soils are therefore a major actor of the global CH<sub>4</sub> cycle. New trends in atmospheric CH<sub>4</sub> studies deal with modelling the retroactive effect of global warming and atmospheric CO<sub>2</sub> increase on CH<sub>4</sub> emissions by terrestrial environments [138] with a special focus on ricefields [157, 183].

The role of soils as source and sink has been discussed before 1997 in general reviews [37, 145, 211], or in reviews dealing with specific environments such as ricefields [147, 230], forests and temperate cultivated soils [203]. Since then, in relation with the increasing scientific and political interest in greenhouse gases, numerous papers have been published on this topic. This review summarises current knowledge with emphasis on recent developments.

## 2. MECHANISMS AND MICROFLORA INVOLVED

### 2.1. Methanogenesis

The complete mineralisation of organic matter in anaerobic environments where sulphate and nitrate concentrations are low occurs through methanogenic fermentation, which produces CH<sub>4</sub> and CO<sub>2</sub> according to the reaction: C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> → 3 CO<sub>2</sub> + 3 CH<sub>4</sub>.

This transformation requires successive actions of four populations of micro-organisms that degrades complex molecules in simpler compounds:

– hydrolysis of biological polymers into monomers (glucides, fatty acids, amino acids) by an hydrolytic microflora that can be either aerobic, or facultatively, or strictly anaerobic;

**Table I.** Soil contribution to atmospheric CH<sub>4</sub> (Tg·year<sup>-1</sup>) according to IPCC [87].

	Estimate	Uncertainty
Sources		
Submerged soils	115	55–150
Other natural sources	50	25–140
Ricefields	60	20–100
Enteric fermentation and animal waste	105	85–130
Energy production and use	100	70–120
Landfills	30	20–70
Biomass burning	40	20–80
Domestic sewage	25	
Total of sources	525	
Sinks		
Consumption in atmosphere	470	420–520
Oxidation in upland soils	30	15–45
Total of sinks	500	

**Table II.** Characteristics of trophic and morphological groups of methanogens.

Trophic groups and substrates	Cocci	Rods	Rods with a sheath*	Sarcinae
Hydrogenotrophs				
H <sub>2</sub> + CO <sub>2</sub>	most	most	none	few
Formatotrophs (all formatotrophs are hydrogenotrophs)				
Formate	several	several	none	none
Acetotrophs				
Acetate	2 species	none	1 genus	all
Methylotrophs				
Methylated compounds	4 genera	none	none	all
Alcoholotrophs (no strict forms)				
Alcohols I, II	none	few	none	few

\* Genus *Methanosaeta*.

– acidogenesis from monomeric compounds and intermediary compounds formed during fermentation (production of volatile fatty acids, organic acids, alcohols, H<sub>2</sub> and CO<sub>2</sub>) by a fermentative microflora that can be either facultatively or strictly anaerobic;  
 – acetogenesis from the previous metabolites by a syntrophic or homoacetogenic microflora; and  
 – methanogenesis from the simple compounds that can be used by methanogens (in particular H<sub>2</sub> + CO<sub>2</sub> and acetate) which constitutes the last step of the methanogenic fermentation.

Methanogenesis, which requires strict anaerobiosis and low oxido-reduction potentials ( $E_h < -200$  mV), involves a specialised, strictly anaerobic microflora that can develop in synergy or in syntrophy with other anaerobic bacteria. Methanogens belong to the domain *Archaea* [242]. A review by Boone et al. [20] presents a detailed taxonomic treatment of methanogens based on the percentage of DNA/DNA hybridisation and the differences in sequences of the rRNA 16S gene. Currently, twenty-six genera and more than sixty species of methanogens have been recorded [67].

Methanogens have a limited trophic spectra comprised of a small number of simple substrates: H<sub>2</sub> + CO<sub>2</sub>, acetate, formate, methylated compounds (methanol, methylamines, dimethylsulphur), and primary and secondary alcohols. This allows to distinguish five trophic groups of methanogens (*table II*). CO can be used by methanogens but is not an important substrate. The two major pathways of CH<sub>4</sub> production in most environments where organic matter decomposition is significant (digesters, freshwater sediments, submerged soils) are acetotrophy and CO<sub>2</sub> reduction by H<sub>2</sub> [41, 185, 205].

Despite the low number of methanogenic species that can use acetate as C and energy source (14 %, corresponding to the genera *Methanosarcina* and *Methanosaeta*), acetotrophy is generally considered responsible for about two-thirds of the CH<sub>4</sub> produced. This reaction produces little energy in normalised conditions ( $\Delta G'_0$ ), which results in a low growth rate of acetotrophic methanogens.

In ricefield soil, H<sub>2</sub> + CO<sub>2</sub>-dependent methanogenesis contributed about 25–30 % of the CH<sub>4</sub> produced,

as shown by CH<sub>3</sub>F acetoclastic inhibition [38], and seemed to be driven by the decay and fermentation of root material [38, 39]. The H<sub>2</sub> inter-species transfer between fermentative and methanogenic bacteria is an important process in anaerobic fermentation. It detoxifies the medium by maintaining the H<sub>2</sub> partial pressure at a low level. About 77 % of methanogenic species are hydrogenotrophic; about 60 % also utilise formate. Formate, like H<sub>2</sub>, is involved in inter-species transfers during the oxidation of the reduced compounds produced by the anaerobic decomposition of organic matter [57, 207]. The energy produced by these reactions is high, which results in a rapid growth of hydrogenotrophic and formatotrophic methanogens.

Methylated compounds are used by about 28 % of methanogens in specific environments such as marine sediments. Methanol, which mostly originates from pectin degradation during the decomposition of algae [180], could be a significant substrate for methanogens in ricefields where large biomasses of freshwater algae often develop [170]. However, the relative contribution of methanol to CH<sub>4</sub> production in submerged ecosystems has not yet been determined.

Methanogenic bacteria have been mostly studied in ricefield soils but they are probably ubiquitous in soils. Waterlogging upland soils, such as forest and cultivated soils, initiated methanogenesis and increased methanogenic populations [135].

Whereas twenty-six methanogenic genera have been currently described, strains isolated or evidenced in ricefield soils belong only to seven genera: *Methanobacterium*, *Methanosarcina*, *Methanobrevibacter*, *Methanoculleus*, *Methanogenium*, *Methanosaeta* and *Methanospirillum* [6, 7, 63, 93, 94, 117, 139, 163, 164]. The characterisation of *Archaea* populations in an Italian rice soil by molecular methods demonstrated the presence of representatives of *Methanosarcina*, *Methanosaeta*, *Methanobacter* and *Methanomicrobium*, together with euryarchaeotal and crenarchaeotal clusters [132]. The same study evidenced (i) clusters including known methanogens and uncultivable strains, and three new clusters of non-cultivable strains, all belonging to *Euryarchaeota* and (ii) two new clusters of *Crenarchaeota* [132]. The number of

non-cultivable strains belonging to clusters of known methanogens or new clusters inserted between clusters including known methanogens were much more numerous than cultivable identified strains. Assuming that most non-cultivable *Archaea* phylogenetically close to known methanogens are methanogens – as supported by the results of a study using cultivation and molecular methods [71] – it can be inferred that a very significant proportion of methanogens present in rice soils are currently uncultivable. Micro-organisms that have been isolated and most studied are not necessarily those that are the only or the most active in soil.

Very little data are available on methanogen abundance in soils other than ricefield soils. Values recorded in twenty-nine ricefield soils in Senegal ranged from  $10^2$  to  $10^7$   $g^{-1}$  dry soil [68]. Apparently populations of cultivable methanogens show little variations during the rice crop cycle. In Italian ricefields, methanogen populations present in the soil before submersion (about  $10^4$  acetotrophs and  $10^6$  hydrogenotrophs  $g^{-1}$  dry soil) were abundant enough to initiate  $CH_4$  production after 100 h submersion without an increase of the population density [135]. In Japanese ricefields, populations of hydrogenotrophs ( $10^3$ – $10^4$ ), methylotrophs ( $10^4$ – $10^5$ ) and acetotrophs ( $10^4$ – $10^5$ ) remained approximately constant during 2 years cropping and were independent from the water management, the type of culture (rice or wheat), fertiliser application and sampling depth (0–1, 1–10 and 10–20 cm) [6]. Molecular methods confirmed the relative stability of *Archaea* in a ricefield soil studied for 17 d after flooding. *Methanomicrobiaceae* and *Methanosaetaceae* did not change in relative frequency. *Methanobacteriaceae* decreased over time; only the relative abundance of *Methanosarcinaceae* increased, roughly doubling from 15 to 29 % of total archaeal gene frequency within the first 11 d, which was positively correlated to the dynamics of acetate and formate concentrations [132].

## 2.2. Methanotrophy

Two forms of  $CH_4$  oxidation are recognised in soils [12, 13, 75].

The first form, known as ‘high affinity oxidation’, occurs at  $CH_4$  concentrations close to that of the atmosphere (< 12 ppm). This form is apparently ubiquitous in soils that have not been exposed to high  $NH_4^+$  concentrations [210]. It is estimated to contribute 10 % of total  $CH_4$  consumption [211]. The second form of oxidation, known as ‘low affinity oxidation’ occurs at  $CH_4$  concentrations higher than 40 ppm. It is performed by bacteria called methanotrophs [92, 110, 238] and is considered as methanotrophic activity *sensu stricto*.

Bacterial population responsible for ‘high affinity oxidation’ are still largely unknown [12]. The study by denaturing gradient gel electrophoresis (DGGE) of *pmoA* genes from strains originating from forest and upland soils that exhibited high  $CH_4$  consumption

rates showed that their DGGE bands were only distantly related to those of known methanotrophs, which indicated the existence of unknown methanotrophs involved in atmospheric  $CH_4$  consumption [77]. A similar conclusion arises from radioactive fingerprinting of micro-organisms that oxidise atmospheric  $CH_4$  in different soils, which could only characterise strains as ‘an unknown group of the alpha *Proteobacteria*’ [78].

Cultivable methanotrophs responsible for ‘low affinity oxidation’ occur in all soils with a pH higher than 4.4 [210]. Already in the 30s, enrichment in organic matter had been observed in soils surrounding leaking gas pipes [211, 214]. Methane oxidation in methanogenic environments (ricefields, peat soils, landfills, etc.) is a low affinity activity. Methane concentration in the water of the first centimetres of a ricefield soil may reach 110 ppm [40] and that in the air of a drained rice soil is often higher than the 11–45 ppm threshold established for a low affinity activity [12].

In wetlands, methanotrophs develop in the oxidised soil layer, in the aerobic rhizosphere of plants possessing an aerenchyma, and inside the roots and the submerged part of the leaf sheaths of the rice plants [21, 69].

Methanotrophs use  $CH_4$  as only a C and energy source. Oxygen availability is the main factor limiting their activity. However, a partial  $CH_4$  oxidation was reported in marine anoxic sediments [4] and was also suspected of occurring in submerged soils [144].

More than 90 % of the  $CH_4$  produced in the anaerobic environments of ricefields can be reoxidised by methanotrophs in the aerobic zones [66, 158, 179]. Depending on the period of the crop cycle and the water management, the percentage of the  $CH_4$  produced that is oxidised by methanotrophs varies from 0 to 97. In a Texas ricefield, during maximum  $CH_4$  production, under continuous irrigation, about 70 % of the  $CH_4$  produced was reoxidised [179].

In ricefields, variations in  $CH_4$  emission were mostly attributed to variations in methanotrophic activity [177, 185]. Similarly, in Florida swamps, an increase in  $CH_4$  emission associated with a decrease of the environmental oxidation was not due to methanogenesis stimulation but to a decrease of the methanotrophic activity [110].

About 80 % of the  $CH_4$  diffusing through the oxidised soil-water interface in ricefields is consumed by methanotrophs [40]. The use of methyl fluoride as methanotrophy inhibitor showed that  $CH_4$  emission would be five to ten times higher in the absence of oxidised soil [10]. But  $CH_4$  oxidation in the rhizosphere is quantitatively the most important and varies according to the development stage of the rice plant [51]. Methanotrophs are also associated with roots and rhizomes of aquatic plants and their activity is correlated with the oxidising activity of the rhizosphere [110]. In oxic soils, maximum methanotrophy is usually observed in the lower soil layer [14].

Methanotroph counts deal mostly with ricefields. Very little data are available, probably because of methodological difficulties [61]. Estimates range from  $10^4 \text{ g}^{-1}$  in Japanese soil [232],  $10^6$  in Italian soils [12],  $8 \cdot 10^5$  in surface soil planted or non-planted, and  $1.4 \cdot 10^6$  in the rhizosphere [21]. Methanotrophs are also associated with the rice plant, with reported densities of  $10^5$ – $10^6 \text{ g}^{-1}$  dry wt of root and  $10^3$ – $10^4 \text{ g}^{-1}$  dry wt in the lower part of culms [232]. Methanotroph abundance increased with the age of the rice plant, whereas it remained approximately constant in submerged unplanted soil [69]. Methanotroph populations in twenty-two dry rice soils ranged from  $10^2$  to  $10^4 \text{ g}^{-1}$  soil dry wt [61]. Populations markedly increased when soils were incubated under an atmosphere enriched with  $\text{CH}_4$  ( $10^7$ – $10^9 \text{ g}^{-1}$  soil dry wt). The enhancement of methanotrophic population by incubation under  $\text{CH}_4$  was observed with soils from ricefields, forests and grasslands [12, 21, 232].

Methanotrophs isolated from ricefields belong to the genera *Methylocystis* [206] and *Methylosinus* [23, 122]. Type II methanotrophs are probably dominant in ricefields because they are the only type producing soluble methane-mono-oxygenase, which avoid the accumulation of  $\text{NO}_2$  toxic to methanotrophs; they also possess resistance forms more efficient than those of type I [107, 122]. Phylogenetic analysis confirmed this hypothesis [53, 76].

Microcosm experiments demonstrated that methanotrophs significantly contributed to nitrification in the rhizosphere, while the contribution of nitrifiers to  $\text{CH}_4$  oxidation was insignificant [18]. This indicate that the beneficial effect of methanotrophs on greenhouse gases balance could be reduced by the production of  $\text{NO}_x$ .

### 2.3. Relations between methanogens and methanotrophs

Counts of methanogens and methanotrophs have been currently performed mostly in ricefield soils. Results show that both groups are ubiquitous in these soils. Dynamic studies seem to indicate that methanogens and methanotrophs maintain their populations under unfavourable conditions, i.e. during drainage and drying-up for anaerobic methanogens and during submersion for aerobic methanotrophs [61, 63, 68, 94, 95]. A study where both populations were simultaneously counted in a range of ricefield soils, confirmed that methanogens and methanotrophs were present simultaneously in ricefield soils and showed that their densities were positively correlated. The densities of cultivable methanotrophs and potential methanotrophic activities were higher than the densities of cultivable methanogens and potential methanogenic activities [93].

### 2.4. Methane transfer from soil to atmosphere

Methane emission by wetland soils results from  $\text{CH}_4$  production in anoxic zones,  $\text{CH}_4$  consumption by

methanotrophs in oxidised zones (rhizosphere, lower part of culms, soil-water interface and submersion water), and transfer to the atmosphere, mostly through rice aerenchyma and, at a lower level, through diffusion and ebullition (*figure 1*).

In planted ricefields, only a low percentage of the  $\text{CH}_4$  produced escapes as bubbles through the soil and the submersion water. Rice plants with their aerenchyma act as pipes, allowing gaseous exchanges between soil and atmosphere [148]. Usually, planted ricefields emit more  $\text{CH}_4$  than wet fallow fields because of a higher C availability for methanogenesis and an easier transfer to the atmosphere, both resulting from the larger aerenchymous plant biomass in planted fields than in fallow fields, as observed by Schütz et al. [185]. Methane emission is a passive transfer through the aerenchyma and micropores located on rice leaves [152]. Methane emission varies with rice varieties [1] probably because of morphological differences in the aerenchyma [27] and root porosity [193]. At the beginning of the crop cycle, when rice plants are little developed, bubble formation and vertical movement in the bulk of the soil is the main transfer mechanism. When rice plants develop, diffusion through the aerenchyma becomes the dominant process, responsible for more than 90 % of the  $\text{CH}_4$  emitted during the reproductive phase of the rice plant [35, 185, 215].

Similarly, in temperate swamps, aquatic plants are responsible for about 90 % of the  $\text{CH}_4$  transfer to the atmosphere [35, 185, 215, 240]. Nycthemeral variations in  $\text{CH}_4$  emission can be related either to stomata opening (*Scirpus* sp.) or to a convection phenomenon related to temperature (*Phragmites* sp.) [218]. Plants possessing an aerenchyma are mostly herbaceous, but also include trees such as *Alnus* which favour  $\text{CH}_4$  emission in wet areas [175].

## 3. METHODS FOR ESTIMATING ACTIVITIES

Methods to estimate  $\text{CH}_4$  production, consumption, and emission by soils should be used with caution, while keeping in mind that they measure complex microbial activities, integrating a larger number of environmental parameters. To be significant, measurements must take into account the spatial and temporal variations as well as the low sensitivity of the methods, especially for  $\text{CH}_4$  measurement at atmospheric concentration [211].

### 3.1. Flux measurements

The closed static chamber is the most frequently used field method to estimate positive (emission) or negative (consumption)  $\text{CH}_4$  fluxes [184]. An alternative used to measure emission is the open chamber, where a continuous gaseous flux is circulated allowing to estimate  $\text{CH}_4$  emission on a given area by difference. The major problem encountered with both methods is the very high variations of measurements in

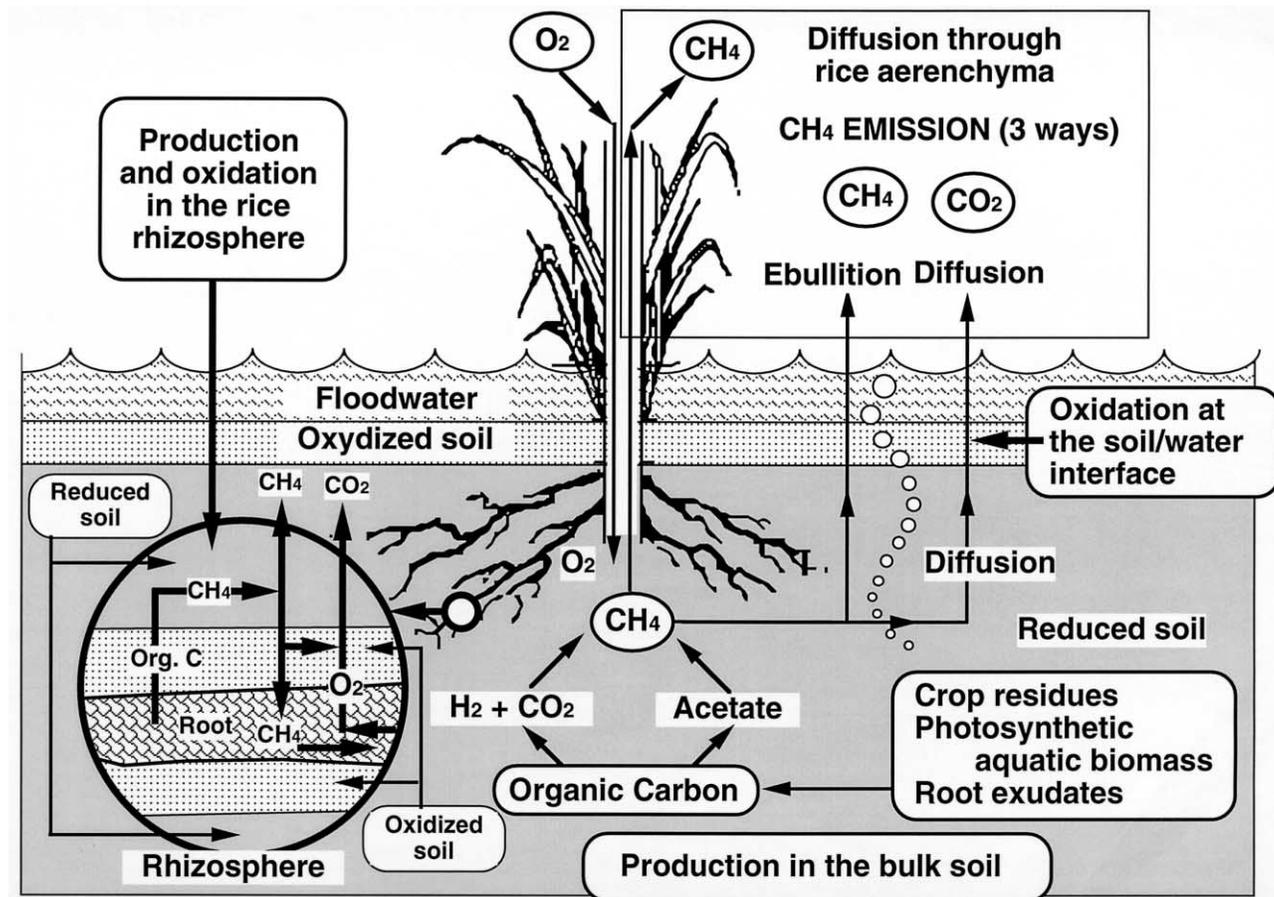


Figure 1. Production, consumption and transfer of CH<sub>4</sub> to the atmosphere in ricefields.

space and time [173]. Soil microbial activity usually exhibit very large variations in space. This was also observed with methanogens [173].

When measuring CH<sub>4</sub> emission, spatial variability is increased by the heterogeneity of the diffusion pathways. Methane emission measurements in Danish soils submitted to temporary flooding showed coefficients of variations ranging from 166 to 1 787 [5]. A bibliographic survey [140] presents 127 estimates of CH<sub>4</sub> emission by ricefields soils submitted to different treatments (36 references). Values range from 0 to 80 mg CH<sub>4</sub>·m<sup>-2</sup>·h<sup>-1</sup>. The study of the data shows a log-normal distribution (coefficient of variation = 94 %) for which the median is more representative than the mean. The median is 9.6 mg CH<sub>4</sub>·m<sup>-2</sup>·h<sup>-1</sup> with a confidence interval at 95 % of -27 and +37 %.

Diurnal and seasonal variations of CH<sub>4</sub> emission can also be very high [191] (see also section 5.2.2). Methane concentration in the air over a ricefield may vary from 23 ppm-vol at the beginning of the day to an average value of 1.75 ppm-vol during the day [188].

Estimating CH<sub>4</sub> emission therefore requires a high number of replicates and integrated measurements at short time intervals. In a study of gaseous emission by

cultivated and forest soils in Canada, between seven and 452 static chambers were needed, depending on the site, to obtain an accuracy of 10 % when measuring gaseous fluxes (CO<sub>2</sub> or CH<sub>4</sub>) higher than 0.15 mg·m<sup>-2</sup>·d<sup>-1</sup> [123].

Various mathematical models have been proposed to estimate CH<sub>4</sub> emission [64] and consumption [70, 168]. They were generally conceived for large scale evaluation and require the input of data that can be very expensive to acquire, such as a combination of Landsat Thematic Mapper, the Advanced Very High Resolution Radiometer satellite images [160] and stable isotope measurements [124]. The design of such models was urged by governments driven by international agreements [176]. They still require a realistic comparison with experimental results [157].

### 3.2. Specific activity measurements

Actual CH<sub>4</sub> production by a soil is generally estimated from soil cores incubated in anaerobiosis. The determination of dissolved CH<sub>4</sub> in soil solution is an alternative non-destructive method, which simplifies the sampling procedure [3]. The measurement of

potential methanogenesis requires the addition of various carbon substrates [93]. Using labelled substrates allows to follow, *in vitro* and *in situ*,  $\text{CH}_4$  production from various substrates. Using this technique allowed to test an anaerobic digestion model used to describe biphasic kinetics of  $\text{CH}_4$  formation in tundra soils. It fit the experimental data rather closely and provided the kinetic coefficients of acetogens, and hydro-geotrophic and acetoclastic methanogens [220].

Actual methanotrophy is estimated by using radon as a tracer [58] or inhibiting  $\text{CH}_4$  oxidation by methylfluoride ( $\text{CH}_3\text{F}$ ). The methanotrophic potential of a soil is estimated by incubation in a close device under an atmosphere enriched with  $\text{CH}_4$  (20 % v/v) [122]. Interpreting potential methanotrophic activity data requires to know the incubation conditions because pre-incubating a soil under an atmosphere enriched with  $\text{CH}_4$  induces an exponential increase of the activity, as compared with fresh soil, when the soil is exposed to  $\text{CH}_4$  concentrations higher than 1 % for more than 12 h [146]. In rice soils submerged after a dry fallow, the ratios between the activities of soils pre-incubated under  $\text{CH}_4$  concentrations higher than a few ppm (low affinity) and those pre-incubated under lower concentrations (high affinity) ranged from 10 to 200 [12]. This indicates that soil methanotrophs are more often in a resting cell status than in a stage of maximum activity [122, 133].

Combining methods allowed to estimate simultaneously  $\text{CH}_4$  production and consumption and to elucidate the functioning of the ecosystem. Inhibiting  $\text{CH}_4$  oxidation by methylfluoride ( $\text{CH}_3\text{F}$ ), has been used *in situ* to quantify  $\text{CH}_4$  production and consumption and has shown that methanotrophs may sometimes consume more than 90 % of the  $\text{CH}_4$  available in submerged soils [158]. An experiment utilising simultaneously seven methods and four taiga soils [237] showed that (i) at atmospheric concentration of  $\text{CH}_4$ , oxidation rates were lower than  $2 \text{ mg CH}_4 \cdot \text{m}^{-2} \cdot \text{d}^{-1}$ , (ii) oxidation occurred in the sixty first centimetres of the soil, (iii) oxidation was maximum in the 10–20 cm zone, (iv) oxidation occurred at  $\text{CH}_4$  concentrations lower than 0.9 ppm, (v) 60 % of  $\text{CH}_4$  was oxidised into  $\text{CO}_2$  and 40 % was incorporated into the biomass, and (vi) exposure to high  $\text{CH}_4$  concentrations induced a methanotrophic activity reaching  $867 \text{ mg CH}_4 \cdot \text{m}^{-2} \cdot \text{d}^{-1}$ .

The kinetic isotope effect for carbon within recent unconsolidated sediments and soils, used in geological studies, allowed to differentiate the various bacterial  $\text{CH}_4$  generation and consumption pathways, and elucidated the cycling of labile sedimentary carbon, including a possible anaerobic oxidation [136].

#### 4. ESTIMATION OF ACTIVITIES IN VARIOUS ENVIRONMENTS

As already indicated, flux and activity measurements present a very large variability that may partly refrain interpretation or comparisons. However, using

large sets of data allows to draw general conclusions. Estimates presented in this section are from a data base we established from 57 references presenting individual or aggregated data [56, 58, 73, 79, 80, 84, 86, 90, 93, 100, 103, 105, 110, 115, 116, 120, 123, 126, 133, 137, 150, 153, 156, 161, 162, 165, 178, 179, 181, 185, 189, 192, 193, 197, 200, 201, 204, 224, 230, 238, 240, 244, 245]. Twenty-eight different units have been used by authors to express fluxes and activities. In order to allow rough comparisons and analysis of the data, values have been converted in  $\text{g CH}_4 \cdot \text{ha}^{-1} \cdot \text{d}^{-1}$  on the basis of 1 200 t soil per hectare and a constant activity during the day.

##### 4.1. Methanogenesis

Data of  $\text{CH}_4$  production by soils, mostly obtained from small samples of ricefields soils incubated in anaerobiosis ( $n = 45$ ) range from 0 to  $78 \text{ kg CH}_4 \cdot \text{ha}^{-1} \cdot \text{d}^{-1}$ . In rice soils enriched with straw ( $n = 22$ ), values may reach  $128 \text{ kg CH}_4 \cdot \text{ha}^{-1} \cdot \text{d}^{-1}$ . Data dealing with swamps and peat soils ( $n = 5$ ) range from 0 to  $50 \text{ kg CH}_4 \cdot \text{ha}^{-1} \cdot \text{d}^{-1}$ . The larger range of  $\text{CH}_4$  production in rice soils, as compared to uncultivated soils, can at least partly be attributed to an usually higher content in easily mineralisable carbon.

##### 4.2. Methanotrophy

Values are distributed within two large groups. Those corresponding to methanotrophy of high affinity, measured in upland soils, range from 0 to  $1.7 \text{ kg CH}_4 \cdot \text{ha}^{-1} \cdot \text{d}^{-1}$ . Higher values were obtained in forest soils, lower values were obtained in cultivated soils (*table III*). Among upland soils, forest soils are probably the most efficient  $\text{CH}_4$  sink. Their higher methanotrophic activity may partly be attributed to a stimulation by a significant methanogenic activity of the litters [99, 187, 202]. Methane concentration in the ten first centimetres of forest soils in New York state was about 500 ppm [249]. Atmospheric  $\text{CH}_4$  oxidation also occurs in extreme environments such as deserts and glaciers, in the floodwater of submerged soils and in the water of the rivers. However, less than 2 % of the  $\text{CH}_4$  produced in river sediments is reoxidised in water [250].

Values corresponding to methanotrophic activities of low affinity measured either *in situ* in methanogenic environment, or in soil samples incubated under an atmosphere enriched with  $\text{CH}_4$ , range from 0 to  $1.7 \text{ t} \cdot \text{ha}^{-1} \cdot \text{d}^{-1}$ . The extremely high maximum value was observed in a sandy soil at the top of a landfill [105].

Results confirm the positive correlation between methanotrophy and methanogenesis, the highest methanotrophic activities being observed in methanogenic environments.

**Table III.** Methanotrophy in various soil types (g CH<sub>4</sub>·ha<sup>-1</sup>·d<sup>-1</sup>).

Environment	No. of data	Minimum	Maximum	Median
Cultivated soils	13	0.00	866	5.5
Grassland soils	7	1.75	485	6.5
Non-cultivated upland soils	6	0.10	228	8.3
Forest soils	17	0.16	1 659	9.9
Wetland soils	9	0	7·10 <sup>5</sup>	172
Upper soil layer in covered landfills	3	7·10 <sup>4</sup>	1.7·10 <sup>6</sup>	4.5·10 <sup>5</sup>

### 4.3. Methane emission

Methane emission in unplanted upland soils temporarily submerged is around a few g·ha<sup>-1</sup>·d<sup>-1</sup>. In submerged soils, highest emissions (median: 3 kg CH<sub>4</sub>·ha<sup>-1</sup>·d<sup>-1</sup>) are observed in ricefields, where the plant biomass provides substrates for methanogenesis and favours CH<sub>4</sub> transfer to the atmosphere, and in freshwater ecosystem without vegetation, which leads to a low methanotrophic activity and high emission by ebullition. In swamps, the median is 700 g·ha<sup>-1</sup>·d<sup>-1</sup>. Emissions are lower in acidic peat bogs (median: 433 g·ha<sup>-1</sup>·d<sup>-1</sup>) (table IV).

## 5. ENVIRONMENTAL FACTORS THAT AFFECT METHANE EMISSION

Factors that affect CH<sub>4</sub> emission by soils are those that affect:

- gas diffusion in relation with the oxydo-reduction level and CH<sub>4</sub> transfer, in particular the water content, the nature of the clays and the type of vegetation;
- microbial activities in general: temperature, pH, Eh, substrate availability, physicochemical properties of soils, etc.;
- methanogenesis and in particular the competition with denitrification and sulphate-reduction;
- methane-mono-oxygenase activity: content in H<sub>2</sub>, CH<sub>4</sub>, ammonium, nitrate, nitrite, and Cu, etc.

Competition and predation may probably affect methanogenic and methanotrophic populations but have not yet been studied [134].

### 5.1. Physicochemical properties of soils

Little data is available on correlations between soil physicochemical properties and CH<sub>4</sub> emission. They

mostly deal with potential methanogenic and methanotrophic activities in ricefield soils for which major physicochemical properties were presented [93, 150, 227]. Results show that interactions are often complex. The study of twenty-nine rice soils in Senegal showed negative correlations between (i) methanogenic potential and (ii) soil conductivity, chlorine content, clay content and C/N ratio [68]. In a study of sixty ricefield soils, no correlation was found between CH<sub>4</sub> production and soil pH in aerobiosis, N-content, organic matter content, soluble C content and cation exchange capacity [227]. Significant correlations may be obtained when grouping soils according to their level of CH<sub>4</sub> production in the absence of organic manuring [227]. Principal component analysis of methanotroph and methanogen counts, potential methanogenic and methanotrophic activities and physicochemical properties in twenty-two ricefield soils [93] indicated that (i) the ratio between potential methanotrophy and potential methanogenesis was mostly governed by methanotrophy, (ii) soils prone to methanotrophy were above neutrality, rich in available P and had a lower clay content, (iii) soil content in active Mn was positively correlated with methanogen and methanotroph densities, and (iv) no correlation was found between soil texture and populations or potential activities [93].

#### 5.1.1. Water content

Soil submersion allows the development of the methanogenic activity and reduces methanotrophic activity by reducing the size of the oxidised zones.

In wet zones in northern USA, variations in CH<sub>4</sub> emission were related with the depth of the water table and the abundance of rooted plants with aerenchyma, both factors being correlated [112, 189]. Laboratory studies with soil cores from swamps and peat bogs

**Table IV.** Methane emission in different soil types (g CH<sub>4</sub>·ha<sup>-1</sup>·d<sup>-1</sup>).

Environments	No. of data	Minimum	Maximum	Median
Upland soils temporarily submerged	5	0	216	3
Freshwater environments without plants	5	0	10·10 <sup>3</sup>	3·10 <sup>3</sup>
Swamps	11	0	17·10 <sup>3</sup>	720
Peatlands	4	6	2·10 <sup>3</sup>	433
Ricefields	23	1	29·10 <sup>3</sup>	10 <sup>3</sup>

showed that CH<sub>4</sub> emission exhibited a negative logarithmic correlation with the depth of the water table (0 to –60 cm) whereas CO<sub>2</sub> emission exhibited a positive linear correlation with this depth [143]. Upland soils, when temporarily submerged, may become CH<sub>4</sub> sources. This was observed in Canada in grasslands and in well drained cultivated soils when snow was melting or during heavy storms in summer [224].

Soil methanotrophic activity is related to its water content. It increases to a value close to field capacity, then decreases when the water content increases [44, 122]. In poorly drained soils around Canadian forests, CH<sub>4</sub> consumption was negatively correlated with soil water content [123]. In Massachusetts forests, a negative correlation was observed between methanotrophy and soil water content when 60 to 100 % of soil porosity was filled and gaseous transfer was reduced; at low water content (22 to 60 %), methanotrophy depended upon soil fertility and was two to three times higher in most fertile soils [32]. In Norwegian forest soils, a small increase in the water content over the field capacity markedly reduced methanotrophy [198].

Methanotrophs remain viable in anaerobiosis and, in the absence of carbon source, are more preserved in anaerobiosis than in aerobiosis. This explains why soils submitted to alternate desiccation and submersion maintain a high methanotrophic potential when environmental conditions allow its expression [171, 172]. In Danish soils temporarily submerged, CH<sub>4</sub> emission and consumption were maximum during the drying-up of the soil, probably because of increased CH<sub>4</sub> diffusion and oxygenation of the soil [5].

### 5.1.2. Oxygen availability and soil Eh

In methanogenic environments, O<sub>2</sub> availability is the major factor limiting methanotrophy. Methanotrophs are ubiquitous in ricefield soils, where their densities were not strongly affected by the oxidation status of the soil [93]. In ricefields, CH<sub>4</sub> oxidation was higher in the rhizosphere followed by surface soil (0.1 cm) and the bulk of the soil (10–20 cm) [118]. The importance of O<sub>2</sub> availability was also evidenced in Florida swamps where methanotrophy was significant in peat, where gas diffusion is easy, whereas methanotrophy was negligible in compact clay soils [110]. In submerged soils and freshwater ecosystems, light availability, which allows benthic photosynthetic activity, increases the thickness of the oxidised soil layer and thus CH<sub>4</sub> oxidation.

Laboratory experiments with submerged soils planted with *Spartina patens* and rice and maintained at Eh values of 100, 0, –100, and –200 mV show that Eh affects not only methanogenesis but also gas transfer through the plant [113]. At lower Eh, aerenchyma formation increased and the size of the roots decreased. A decrease in Eh from –200 to –300 mV induced a ten-fold increase in CH<sub>4</sub> production and a 17-fold increase in its emission [114].

### 5.1.3. Organic matter content

The intensity of reduction processes in submerged soils depends upon the content and nature of organic matter (OM), the ability of the microflora to decompose this OM, and the availability and nature of electron acceptors. The Eh of ricefield soils rich in active Fe and organic matter may reach values lower than –200 mV in less than 2 weeks [148]. A positive correlation may therefore be observed between the methanogenic potential and the OM content in soils. In five rice soils in Japan, CH<sub>4</sub> emission ranged from 0.6 to 8.2 g CH<sub>4</sub>·m<sup>–2</sup> and was maximum in a peat soil where mineralisable C was highest. However, such a correlation was not invariably observed. The study of twenty-nine ricefield soils comprising eighteen saline soils showed a positive correlation between methanogenesis and OM content only in non-saline soils [68], which was explained by the inhibitory effect of salinity on methanogenesis. Similarly, a positive correlation between CH<sub>4</sub> production and OM content was observed only in soils exhibiting a high methanogenic activity [227].

In peat soils, the nature of the OM determined both CH<sub>4</sub> production and consumption, both activities being correlated [143]. In the wet peat soils, significant CH<sub>4</sub> production only occurred from organic matter fractions with a large particle size; the fraction > 2.0 mm contributed 90 % of the total CH<sub>4</sub> production capacity. Methane production capacity strongly decreased with depth; the layer 0–5 cm contributed 70 % of the total CH<sub>4</sub> produced, indicating that recent plant residues are a major substrate for methanogens [217]. Atmospheric CH<sub>4</sub> consumption by a tundra soil increased four times when glucose was added to the soil [234].

### 5.1.4. pH

The activity of methanogens is usually optimum around neutrality or under slightly alkaline conditions [67] and is very sensitive to variations in soil pH [226]. The minimum pH allowing the growth of 68 methanogenic species was 5.6 [67]. Studies in clay soils in Texas showed that CH<sub>4</sub> emission was four times lower in the most acidic soil with low structural stability [177]. However, methanogens can adapt to acidic environment. Methane production and consumption in peat soils in temperate and subarctic areas (pH 3.5–6.3) was optimum between 5.5 and 7.0 for methanogenesis and between 5.0 and 6.5 for methanotrophy [60]. Methanotrophs are more tolerant to pH variations than methanogens [60]. They are, however sensitive to the acidification of the environment. In non-fertilised permanent grassland at the Rothamsted experimental station, methanotrophy decreased from –67 to –35 nL CH<sub>4</sub>·L<sup>–1</sup>·h<sup>–1</sup> when pH decreased from 6.3 to 5.6 and was fully refrained at lower pH [86]. Molecular ecological methods have however evidenced acidophilic methanotroph, non-cultivable on classical media, in peat soils at pH < 4.7 [133].

### 5.1.5. Soil texture and mineralogy

In submerged soils, texture is involved in (i) the establishment of the anaerobiosis needed for methanogenesis, (ii) protecting organic matter from decomposition, (iii) the transfer and trapping of CH<sub>4</sub> produced in the reduced soil, and (iv) affecting the depth of the oxidised soil layer hosting methanotrophs.

One could expect clay soils, which are poorly drained and prone to anaerobiosis, to favour methanogenesis. The study of 132 ricefield soils in Japan showed CH<sub>4</sub> emissions higher in gley soils than in the other types of soils. On the other hand, a negative correlation was observed between the methanogenic potential and the clay content in ricefield soils in Senegal [68]. More than clay content, the nature of the clay affects CH<sub>4</sub> emission because some clay types protect organic matter from mineralisation [155], which delays methanogenesis. Soils rich in swelling clays are usually more favourable to methanogenesis than sandy soils, silty soils or soils rich in kaolinite, where density increases after submersion, slowing down pH and Eh variations and organic matter decomposition [149]. A field study in two Thailand soils showed that rice straw decomposition was slower in soil rich in kaolinite [149]. However, in Indian ricefields, higher CH<sub>4</sub> emission was reported in inceptisols (8–21 g·m<sup>-2</sup>) than in swelling vertisols (1.5–11 g·m<sup>-2</sup>) [194].

A high clay content can also favour trapping of CH<sub>4</sub> bubbles in soils [178] and decreases emission. In the Philippines, CH<sub>4</sub> emission and production during three crop cycles were markedly higher in a calcareous sandy silt than in a clay soil [52]. In calcareous soils, CH<sub>4</sub> production seems to be partly stimulated by a buffering effect of carbonates [148]. In sandy ricefield soils in Texas, a positive correlation was observed between sand content (18.8 to 32.5 %) and the average CH<sub>4</sub> emission during the crop (15.1 to 36.3 g·m<sup>-2</sup>) [178].

Methane production in different textured model soils demonstrated that a high amount of negative surface charges increased CH<sub>4</sub> production under both oxic and anoxic conditions. Methane production rates in marshland soils increased in the following order: sand < gravel < clayey silt < clay. Indigenous microflora in combination with the sorptive quality of soil particles (clay, silt, organic matter) enabled methanogenic activity in the presence of oxygen, promoting micro-scale anoxia within the slurries [221].

### 5.1.6. Chemical properties

Nitrate reducers, ferric iron reducers and sulphate reducers constitute a sequence of competitors of methanogens for acetate and electrons [33]. A high Fe content of the soils, which allows a fast Eh decrease after submersion, favours methanogenesis [93, 226]. Ferric iron can have both a chemical impact, because of its reoxidation by root O<sub>2</sub>, and a biological impact, by increasing C oxidation into CO<sub>2</sub> [65, 248]. Fe(III) may reduce CH<sub>4</sub> production in ricefield soils by

maintaining the activity of the micro-organism implied in its reduction, thus delaying substrate availability for methanogenic bacteria [230].

Methane emission is usually lower in sulphate [245] and acid-sulphate soils [91] than in the other soil types. This mainly results from the competition for H<sub>2</sub> between methanogens and sulphate reducers, but the lower rice productivity in sulphate soils may also contribute to the decrease in CH<sub>4</sub> production. In Thailand, CH<sub>4</sub> emission was ten times lower in sulphate soils (2–4 mg·m<sup>-2</sup>·h<sup>-1</sup>) than in non-sulphate soils (20–30 mg·m<sup>-2</sup>·h<sup>-1</sup>) [245]. However, thermodynamical experiments using sixteen ricefields soils showed that CH<sub>4</sub> production depended mainly on the availability of degradable organic substrates rather than the amount of reducible sulphate and ferric iron [247, 248].

Experiments in ricefields showed that an artificial increase in salinity (0.66 kg·m<sup>-2</sup>) decreased CH<sub>4</sub> production by three to four times and its emission by 25 %, which indicated a higher sensitivity of methanotrophy to salinity as compared to methanogenesis [50]. Phosphorus addition on planted rice soils significantly decreased CH<sub>4</sub> emission [131] probably by increasing methanotrophic potential [95]. Heavy metal impact on CH<sub>4</sub> production is complex and generally seems to be inhibitory [142].

## 5.2. Climatic factors

### 5.2.1. Temperature

Methanogenesis is optimum between 30 and 40 °C. Low soil temperatures reduce CH<sub>4</sub> production by decreasing the activity of methanogens but also that of other bacteria implied in methanogenic fermentation. The latter seems to be more sensitive than methanogens to temperature variations [42]. Variations in CH<sub>4</sub> production in waterlogged soils in relation with temperature may partly be due to a variation of Q<sub>10</sub> (relative increase in activity after an increase in temperature of 10 °C) in time for methanogens [219] and different values of Q<sub>10</sub> for populations which compete with methanogens for H<sub>2</sub>: 2.4 for iron reduction, 1.6 for sulphate reduction, and 4.6 for CH<sub>4</sub> production [216].

Laboratory studies with soil cores from swamps and peatlands in Canada showed that CH<sub>4</sub> emission increased by 6.6 times when incubation temperature increased from 10 to 23 °C [143]. In temperate or cold regions, seasonal variations of CH<sub>4</sub> emission were correlated with soil temperature [112]. These were also observed in subtropical zone [19, 161]. However, significant CH<sub>4</sub> emissions were still observed in swamps in winter. Emission rates of 3 to 49 mg CH<sub>4</sub>·m<sup>-2</sup>·d<sup>-1</sup> were measured in various wetlands under the snow in northern Minnesota [54].

Methanotrophy seems to be less sensitive to temperature than methanogenesis. Methane production and consumption in temperate and subarctic peats was optimum around 20–30 °C for both activities, with a

broader tolerance for methanotrophy than for methanogenesis [60]. Methanotrophy by soil cores from temperate forest did not show large variation between  $-1$  and  $30$  °C [108]. Observation in Massachusetts forests showed that methanotrophy was affected between  $-5$  and  $10$  °C but not between  $10$  and  $20$  °C [31]. Significant methanotrophy was still observed in forest soil in Norway at average temperatures lower than  $1$  °C [198].

Temperature also affect  $\text{CH}_4$  transport through the rice plant [152] as shown by a positive correlation between soil temperature at  $-5$  cm and plant conductance for  $\text{CH}_4$  [81].

Daily variations of  $\text{CH}_4$  emission in ricefields were related with temperature variations during the day [178, 186]. In the Philippines, they followed a consistent pattern, with the highest rates observed in the early afternoon and lowest rates in the early morning [229]. Nycthemeral variations showed emission peaks at night [223] that could be due to a lower activity of methanotrophic bacteria when  $\text{O}_2$  was limiting ( $< 2$  ppm) because no photosynthetic activity occurred [170].

### 5.2.2. Seasonal variations

In methanogenic environments of temperate regions, seasonal variations of  $\text{CH}_4$  emission were related to temperature and insolation [201]. At high latitudes such variations were especially marked [137]. In river sediments in Australia,  $\text{CH}_4$  emission ranged from  $< 0.01$  in winter to  $2.75 \text{ mmol}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$  in summer [19]. However, in temperate ricefields, variation in  $\text{CH}_4$  emission during the crop cycle did not correlate with soil temperature whereas nycthemeral variation did [186].

In addition to a direct effect of the temperature, seasonal variation of  $\text{CH}_4$  emission by wetlands in temperate climate was also related to the vegetative cycles of plants possessing an aerenchyma, and non-rooted floating vegetation, which may play an important role in  $\text{CH}_4$  oxidation, as observed in North Carolina swamps [100].

## 5.3. Role of the vegetation in submerged soils

### 5.3.1. In ricefields

The presence of rice strongly increased  $\text{CH}_4$  emission by providing C sources [46] and by favouring  $\text{CH}_4$  transfer to the atmosphere. In a Louisiana soil,  $\text{CH}_4$  emission in 77 d was  $50 \text{ kg}\cdot\text{ha}^{-1}$  in unplanted control and  $220 \text{ kg}\cdot\text{ha}^{-1}$  in planted field [126]. The quantity of  $\text{CH}_4$  emitted during the crop cycle was positively correlated ( $r^2 = 0.845$ ,  $n = 11$ ) with the aerial vegetative biomass of rice. Daily production was correlated with the aerial biomass ( $r^2 = 0.887$ ,  $n = 93$ ) and root biomass ( $r^2 = 0.816$ ,  $n = 33$ ). Carbon emitted as  $\text{CH}_4$  corresponded to 3 and 4.5 % of the photosynthetic carbon in rice varieties with low or high potential for  $\text{CH}_4$  emission, respectively [82]. All yield parameters, including the number of tillers, were

correlated with  $\text{CH}_4$  emission [193]. As rice yield is usually higher during the dry season than during the rainy season,  $\text{CH}_4$  emission is higher during the dry season. In the Philippines, a rice yield of  $5.2\text{--}6.3 \text{ t}\cdot\text{ha}^{-1}$  during the dry season corresponded to an average emission of  $190 \text{ mg}\cdot\text{CH}_4\cdot\text{m}^{-2}\cdot\text{d}^{-1}$  and a yield of  $2.4\text{--}3.3 \text{ t}\cdot\text{ha}^{-1}$  during the wet season to  $79 \text{ mg}\cdot\text{CH}_4\cdot\text{m}^{-2}\cdot\text{d}^{-1}$  [229].

### 5.3.2. In swamps

In swamps, plants with an aerenchyma favour  $\text{CH}_4$  emission by allowing its transfer to the atmosphere [200], whereas plants without aerenchyma reduce its emission, partly because of rhizospheric oxidation. In swamp areas with no vegetation,  $\text{CH}_4$  emission by ebullition was higher than that in areas with rooted plants; in the presence of rooted plants, the percentage of  $\text{CH}_4$  in the biogas was lower (42–45 %) than in the bare areas (60 %), which confirmed the major role of the rhizosphere in  $\text{CH}_4$  oxidation [201]. Methane emission in Hudson Bay peatlands was three to thirty times higher in areas with no vegetation than in the adjacent zones colonised by plants [73].

In Michigan peatlands with a water table at  $-20$  cm, a slight  $\text{CH}_4$  consumption ( $-0.2$  to  $-1.5 \text{ mg}\cdot\text{CH}_4\cdot\text{m}^{-2}\cdot\text{d}^{-1}$ ) was observed in bush zones, whereas emission was observed in areas colonised by plants with aerenchyma [189]. Similarly, in wet tundra soils, no  $\text{CH}_4$  emission was observed in the absence of vascular vegetation, indicating that in the absence of  $\text{CH}_4$  transport through plants, the upper layer of such soils behaved as an efficient biofilter [111, 212].

Usually,  $\text{CH}_4$  emission and the net productivity of cultivated or non-cultivated submerged soils are positively correlated. About 3 % of the daily productivity is emitted as  $\text{CH}_4$ . The increase in atmospheric  $\text{CO}_2$ , which increases ecosystem productivity should also increase  $\text{CH}_4$  emission by wetlands [45, 239].

Variations of  $\text{CH}_4$  flux in a given area might appear as related to vegetation, but relationships are often more complex. In an alpine tundra ecosystem [235],  $\text{CH}_4$  production was observed in *Carex* sites. A significantly higher value (seasonal mean:  $+8.45 \text{ mg}\cdot\text{CH}_4\cdot\text{m}^{-2}\cdot\text{d}^{-1}$ ) in one site as compared with other similar sites (seasonal means:  $-0.06$  and  $+0.05 \text{ mg}\cdot\text{CH}_4\cdot\text{m}^{-2}\cdot\text{d}^{-1}$ ) was attributed to a shallower snowpack during winter. In *Acomastylis* meadows, which had an intermediate moisture regime,  $\text{CH}_4$  oxidation dominated (seasonal mean:  $0.43 \text{ mg}\cdot\text{CH}_4\cdot\text{m}^{-2}\cdot\text{d}^{-1}$ ). In a windswept *Kobresia* meadow plant community, which received the least amount of moisture from snowmelt, only  $\text{CH}_4$  oxidation was observed (seasonal mean:  $-0.77 \text{ mg}\cdot\text{CH}_4\cdot\text{m}^{-2}\cdot\text{d}^{-1}$ ). Methane fluxes correlated with a different set of environmental factors within each plant community. In *Carex* plant community,  $\text{CH}_4$  emission was limited by soil temperature. In *Acomastylis* meadows,  $\text{CH}_4$  oxidation rates correlated positively with soil temperature and negatively with soil moisture. In the *Kobresia* community,  $\text{CH}_4$  oxidation was stimulated by precipitation [235].

#### 5.4. Role of soil fauna

In waterlogged ricefields, soil and water fauna, especially aquatic *Oligochaetes*, significantly affect soil texture and Eh, and microbiological activities [170]. One could expect a resulting effect on CH<sub>4</sub> emission but no data is available. A study of the effect of chironomid larvae on CH<sub>4</sub> production, oxidation, and fluxes in a rice soil concluded that they had no effect on CH<sub>4</sub> flux across the sediment surface either by diffusion or by ebullition, but that chironomid tubes were microsites with an intensified microbial activity, where CH<sub>4</sub> production and oxidation might be tightly coupled [97].

### 6. EFFECTS OF CULTURAL PRACTICES IN WETLANDS

Cultural practices in wetlands (planted mostly with rice but also with other aquatic plants such as jute or waterchestnut) affect CH<sub>4</sub> emission through their effects on methanogenesis, methanotrophy and CH<sub>4</sub> transfer.

#### 6.1. Effect of submersion and water management

Rice is mostly cultivated in submerged conditions because wetland rice has a higher yield (up to 10 t·ha<sup>-1</sup>) than upland rice (0.5 to 4 t·ha<sup>-1</sup>) [47]. The physicochemical processes and successive steps of the establishment of anaerobiosis in ricefield soils that allow methanogenesis after submersion are well known [148]. When the soil is submerged, dissolved O<sub>2</sub> concentration rapidly decreases and facultative anaerobic, then microaerophilic, and finally strict anaerobic micro-organisms develop. They use successively various electron acceptors for their respiration: NO<sub>3</sub><sup>-</sup> at Eh values lower than 350 mV, Mn<sup>4+</sup> at Eh < 200 mV, Fe<sup>3+</sup> at Eh < 100 mV and SO<sub>4</sub><sup>2-</sup> at Eh around -150 mV [159]. Strict thiosulphate reduction, which was recently shown to be as important as sulphate reduction in rice soils [62] may develop at an Eh between -150 and -200 mV. Thiosulphate being more reduced than sulphate should be the last electron acceptor before methanogenesis. This succession of reductive reactions may rapidly lead to an Eh around -200 mV, favourable to the reduction of CO<sub>2</sub> into CH<sub>4</sub>. Simultaneously, concentrations in CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> increase, which stabilises soil pH around neutrality. The shift of soil pH toward neutrality after submersion is observed for both acidic and alkaline soils [148]. The depth of the submersion water affects anaerobiosis but also water temperature, which in turn affects microbial activities and CH<sub>4</sub> transfer [83].

Numerous in situ studies report a significant decrease (60 to > 90 %) of CH<sub>4</sub> emission by ricefields that are drained one or several times during the crop cycle [26, 98, 141, 151]. In Texas ricefields, average CH<sub>4</sub> emission expressed in mg·m<sup>-2</sup>·d<sup>-1</sup> were 106 for classical continuous irrigation, 56 when the field was

drained in the middle of the crop cycle, 13 when the field was drained three times, and 151 for a late continuous irrigation [179]. However, in Indonesia, CH<sub>4</sub> emissions were reportedly similar in fields continuously submerged or weekly drained [153]. Short drainage induce the formation of sulphate and ferric iron, which allows the development of competition for H<sub>2</sub> between methanogens and sulphate reducers + ferro-reducers, which in turn induces an inhibition of methanogenesis that persists after soil reflooding [166]. Water management between crops is also an important factor. A dry fallow emitted less CH<sub>4</sub> during the next crop cycle than a wet fallow [213]. Increasing water percolation in soil might also reduce CH<sub>4</sub> emission [230] but the economic feasibility of this method is doubtful.

A number of water management strategies have been tested to produce more rice with less water [72]. They are based on reduced water depth and reduced time of submersion by maintaining the soil saturated without standing water. For example, a minimum irrigation technology has been developed in China for two decades. It involves the following water management stages: thin water layer at transplanting and during seedling recovery, wet soil without standing water before tillering, drying-up of the soil during tillering (until crack formation on the soil surface), thin water layer from panicle initiation to grain milky stage, and wet soil for the end of the maturation stage. On average, this technique saved 21 % of irrigation water and increased yield by 11.4 %. It has been popularised over 950 000 ha [243]. Such techniques may reduce CH<sub>4</sub> emission but their effect has not yet been determined [230].

#### 6.2. Effect of fertilisers

Rice demand is increasing, whereas the area available to grow rice is already almost fully utilised. Increasing yield requires to increase fertilisation. In most rice producing countries, the use of organic fertiliser has been highly recommended to reduce chemical fertiliser use and maintain soil fertility in the long term. But organic fertilisers (rice straw, green manure or farmyard manure) rich in carbon favour CH<sub>4</sub> production much more than chemical fertilisers do. The nature of fertilisers used, organic or chemical, mainly depend upon the agro-economical conditions, but a right balance should be found to maintain fertility in the long term and mitigate CH<sub>4</sub> emissions.

##### 6.2.1. Organic fertilisers

All in situ studies have shown that organic matter incorporation markedly increased CH<sub>4</sub> emission [26, 98, 126]. As an example, incorporating rice straw (5–12 t·ha<sup>-1</sup>; C/N about 60) increased CH<sub>4</sub> emission by two to nine times in ricefields in Italy [185], Texas [177], Japan [244] and the Philippines [228]. Emission increased linearly with the quantity (0 to 3 %) of straw incorporated [225]. Organic matter incorporation

favoured more CH<sub>4</sub> emission during the dry season when rice biomass is higher, than during the wet season [244]. Methane production and emission decrease when the C content and the C/N ratio of the incorporated material decrease. A high C/N, as in rice straw, usually corresponds to an organic material rich in labile C and thus easily usable by the microflora. Incorporation of *Sesbania* green manure with a lower C/N ratio than straw, increased CH<sub>4</sub> emission by two to five times, according to the quantity incorporated [49, 120]. Incorporating straw compost with low C/N increased CH<sub>4</sub> emission by less than two times. Field experiments showed that the Nouchi model [152] may predict the effect of straw incorporation on CH<sub>4</sub> emission in ricefield soils [22].

Growing *Azolla* (an aquatic fern used as green manure) had a moderating effect on CH<sub>4</sub> efflux from flooded soil that was attributed to an increase in the dissolved oxygen concentration at the soil/floodwater interface [16].

### 6.2.2. Chemical fertilisers

The reported effects of chemical N-fertilisers on CH<sub>4</sub> emission are complex and sometimes contradictory. They depend on the nature of the fertiliser, the quantity applied [125] and the method of application. By increasing rice productivity, fertilisation may increase CH<sub>4</sub> emission. This was observed with urea applied in continuously flooded ricefields [126]. An increase in soil pH resulting from urea hydrolysis might also have favoured CH<sub>4</sub> production [225]. On the other hand, in intermittently drained ricefields, urea application (100 and 300 kg N·ha<sup>-1</sup>) resulted in a 7 and 14 % decrease (respectively) in CH<sub>4</sub> emission as well as in an increase in N<sub>2</sub>O emission [29].

Electron acceptors other than CO<sub>2</sub>, especially nitrate and sulphate, may cause bacterial competition unfavourable to methanogens and decrease CH<sub>4</sub> production/emission. As H<sub>2</sub> and acetate are preferentially used by sulphate-reducing bacteria, sulphate application generally reduces methanogen activity. Ammonium sulphate was frequently reported to significantly (30 to 60 %) reduce CH<sub>4</sub> flux from ricefields [26, 29, 185]. When (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> was applied, the inhibition of CH<sub>4</sub> production was not associated with an increase in soil Eh, which did not change significantly; a direct inhibitory effect of sulphate on methanogenesis was assumed [225]. Despite some contradictory results showing an increase in CH<sub>4</sub> emission after ammonium sulphate application [35], most data support the idea that ammonium sulphate use is a possible way to reduce CH<sub>4</sub> emission from ricefields. Gypsum (calcium sulphate) is used to restore the fertility of saline and/or alkaline rice soils. An additional effect is a significant decrease of CH<sub>4</sub> emission (–29 to –46 % with applications of 1 and 2 t·ha<sup>-1</sup> gypsum [127], 50 to 70 % with application of 6.6 t·ha<sup>-1</sup> [48], 47, 46 and 51 % with applications of 2.5, 5 and 10 t·ha<sup>-1</sup> [130]). This inhibitory effect seems to be independent from the nature of the N-fertiliser

used: the application of 6.7 t·ha<sup>-1</sup> gypsum reduced CH<sub>4</sub> emission by 50 and 70 % in ricefields fertilised with urea or green manure, respectively [48]. However, sulphate addition might be detrimental to rice by favouring rhizospheric sulphate-reduction [47].

Similarly, nitrate application causes a competition for H<sub>2</sub> between denitrifying bacteria and methanogens, that favours denitrifying bacteria. Nitrate is also an oxidant that reduces CH<sub>4</sub> emission by increasing soil Eh [96, 174]. This inhibitory effect was reported to be short-termed: addition of 300 mg·kg<sup>-1</sup> NO<sub>3</sub><sup>-</sup>-N increased soil Eh by 220 mV and almost completely inhibited CH<sub>4</sub> production, but soon after, the applied NO<sub>3</sub><sup>-</sup> was reduced through denitrification and CH<sub>4</sub> production increased [225].

A number of experiments have compared different fertilisers and modes of application. They confirm the advantage of ammonium sulphate which may reduce CH<sub>4</sub> emission by 50–60 % as compared with urea [106, 125]. In Louisiana ricefields, the control without N-fertiliser emitted 60 kg CH<sub>4</sub>·ha<sup>-1</sup>·crop cycle<sup>-1</sup>. Applying 60 kg N·ha<sup>-1</sup> increased emission by 10 kg with ammonium sulphate, 20 kg with potassium nitrate, and 50 kg with urea. Applying 120 kg N·ha<sup>-1</sup> increased emission by 40 kg with ammonium sulphate, 30 kg with potassium nitrate, and 160 kg with urea. Ammonium itself might also indirectly inhibit CH<sub>4</sub> production, nitrates produced by its nitrification being inhibitory for methanogenesis [40, 96]. On the other hand, ammonium being inhibitory for methanotrophy, may reduce CH<sub>4</sub> reoxidation and increase its emission [40].

Methane emission seems to be reduced when N-fertiliser is incorporated, as compared with surface application [185]. In a rain-fed lowland ricefield, deep placement of urea super-granules reduced CH<sub>4</sub> emissions as compared with prilled urea broadcasting [167]. A higher emission when N-fertiliser was surface-applied might be due to an inhibitory effect of ammonium on methanotrophy, which was observed in oxic soils [134] but also at the soil-water interface in a submerged soil [40].

The increased CH<sub>4</sub> emission due to organic manure can be mitigated by combining organic and mineral fertilisation. Ammonium sulphate combined with organic manure reduced emission by 58 % as compared with organic manure alone and increased yield by 32%. Emission peaks were suppressed at tillering and during the reproductive stages of rice [190]. Experiments in Indonesia also showed a significant decrease in CH<sub>4</sub> emission when organic manure was combined with ammonium sulphate and even urea [154].

### 6.3. Effect of pesticides and microbial inhibitors

Little data are available on the impact of pesticides on CH<sub>4</sub> emission under laboratory conditions. Herbicide bromoxynil and insecticide methomyl inhibited CH<sub>4</sub> oxidation in soil slurries [209]. A commercial formulation of fungicide tridemorph to a tropical flooded rice soil stimulated CH<sub>4</sub> production at low levels (5–20 µg·g<sup>-1</sup>) but inhibited the process at

50–100  $\mu\text{g}\cdot\text{g}^{-1}$ . Oxidation of  $\text{CH}_4$  was progressively inhibited with increasing concentrations of the fungicide [15].

Acetylene inhibits nitrification and methanogenesis. When applied as encapsulated calcium carbide, it was slowly liberated and decreased by 90 %  $\text{CH}_4$  emission and  $\text{N}_2\text{O}$  emission, which reduced N-losses [24]. In situ, calcium carbide application reduced  $\text{CH}_4$  emission by 35 % [129] and increased rice yield by 30 % [9]. Dicyandiamide, another nitrification inhibitor, also reduced  $\text{CH}_4$  emission [129].

In both upland and wetland soil, methanotrophy was inhibited by nitrification inhibitors (thiourea, sodium thiosulphate, dicyandiamide, nitrapyrine, calcium carbide). This was attributed to the structural similarity between methanotrophic and nitrifying bacteria. The urease inhibitor N-(*n*-butyl) thiophosphoric triamide (NBPT) also inhibited methanotrophy [25, 118].

Bacterial inhibitors are still at the experimental level and adoption by farmers is strongly refrained by technical and economical constraints. In particular, the application of urease or nitrification inhibitors is not common in rice cultivation. However, as calcium carbide significantly reduces  $\text{CH}_4$  emission and increases rice yield through its inhibitory effect on nitrification, it has a potential for adoption.

#### 6.4. Effect of rice varieties

The increasing demand for rice has led to the selection of rice varieties with a high productivity (high grain/straw ratio), efficient in utilising soil nitrogen, and resistant to parasites and diseases. Some recent studies deal with the selection of varieties with an aerenchyma that reduces  $\text{CH}_4$  transfer.

Varietal differences in  $\text{CH}_4$  emission have been demonstrated. Under continuous irrigation, average emission was 20  $\text{mg}\ \text{CH}_4\cdot\text{m}^{-2}\cdot\text{h}^{-1}$  for IR64 and 14  $\text{mg}\ \text{CH}_4\cdot\text{m}^{-2}\cdot\text{h}^{-1}$  for Cisadane variety [83]. Varietal differences of almost 500 % were observed for Chinese rice varieties [190]. Watanabe et al. [231] have shown that  $\text{CH}_4$  emission differed among varieties but could not evidence correlations with rice type (*japonica*, *indica*), the number of tillers, the height and the biomass of the plant, or the root biomass. Rice variety IR65597 emitted about 30 % less  $\text{CH}_4$  than the traditional variety Dular, whose stems and roots are more developed [151, 222]. Short varieties tested in Louisiana emitted less  $\text{CH}_4$  (185  $\text{kg}\cdot\text{ha}^{-1}$ ) than those with high stems (300  $\text{kg}\cdot\text{ha}^{-1}$ ) [128].

Root exudation, which produces organic substrates directly or indirectly utilised for  $\text{CH}_4$  production, varies qualitatively and quantitatively with rice varieties [119, 135]. This allows to select varieties that produce less exudates and root exfoliations. However, such characteristics reduce the plant's ability to favour associative  $\text{N}_2$ -fixation and efficiently use soil nitrogen [119].

#### 6.5. Dynamics of methane emission during the crop cycle

Methane emission during the rice crop cycle varies with substrate availability, soil Eh, and cultural practices. Different dynamics were reported [150, 178, 196, 245]. Three emission peaks may occur during the crop cycle. The first peak, observed shortly after submersion, is attributed to the decomposition of the easily mineralisable organic matter; it is usually observed in soil where organic manure was applied [150, 231]. A second peak is usually observed during the reproducing phase of the crop and seems to be related to an increased rhizospheric exudation. The incorporation into the soil of the photosynthetic aquatic biomass by weeding and by the activity of the soil fauna also provides organic matter to the soil [170] and may contribute to the anaerobic fermentation process at this stage. A third peak observed at the end of the crop cycle could result from an input of organic matter due to root exfoliation and plant senescence. Variations during the crop cycle cannot generally be attributed to temperature variations [186] and depend mostly on substrate availability for  $\text{CH}_4$  production. A last peak of emission is observed after rice harvest, when discontinuing irrigation leads to the formation of soil cracks through which trapped  $\text{CH}_4$  escapes [28]. This peak was reported to provide about 10 [52] to 20 % [229] of the  $\text{CH}_4$  emitted during the whole cycle.

Dominant activities of the trophic groups of methanogens vary during the crop cycle. Acetotrophy contribution to  $\text{CH}_4$  produced decreased from 67–80 % at the beginning of the crop cycle to 29–60 % during the crop cycle [215].

In most cases,  $\text{CH}_4$  emission by ricefields was higher during the second half of the crop cycle. In California,  $\text{CH}_4$  emission was 5  $\text{g}\ \text{CH}_4\cdot\text{m}^{-2}$  during the two to three last weeks before harvest while average emission during the crop was 0.25  $\text{g}\ \text{CH}_4\cdot\text{m}^{-2}$  [36]. In Texas more than 75 % of total  $\text{CH}_4$  was emitted during the last 5 weeks of the crop cycle [82]. Maximum emission values were observed during flowering [79, 193] and maturation stages [82, 229], which corresponded to about 50 % of the total rice biomass [82]. The percentage of carbon from the photosynthetic production emitted as  $\text{CH}_4$  increased from 0.9–2.0 during the vegetative phase to 3.6–5.0 during the reproductive phase and to 7.9–8.3 during the maturation phase [82]. However a maximum emission was reported during the first half of the crop cycle in three types of ricefield soils in Thailand [91].

### 7. EFFECTS OF CULTURAL PRACTICES IN UPLANDS AND FORESTS

Cultural practices in upland soils mostly affect their potential to oxidise atmospheric  $\text{CH}_4$ . Nitrogen fertilisation that lead directly or indirectly to an increase in the  $\text{NH}_4$  content of the soil has an inhibitory effect on  $\text{CH}_4$  oxidation, through competition at the level of the

methane-mono-oxygenase towards nitrification [30, 146] and the toxicity of  $\text{NO}_2$  produced. Cultural practices that destroy micro-aerophilic niches suitable for  $\text{CH}_4$  oxidisers also reduces atmospheric  $\text{CH}_4$  oxidation [86, 199].

## 7.1. Fertilisers

### 7.1.1. Organic fertilisers

Green manure incorporation (clover residues) in upland cultivated and forest soils in Louisiana reduced methanotrophy by an average 42 % [146]. On the other hand, long-term experiments (140 year) at Rothamsted Experimental station (UK) did not show any inhibitory effect of organic manure on the soil potential to oxidise atmospheric  $\text{CH}_4$  [85]. Most probably, the difference between both treatments was a greater release of ammonium by green manure as compared with farmyard manure with a lower C/N ratio.

### 7.1.2. Chemical fertilisers

In upland soils, the effect of N-fertilisation on soil potential to oxidise atmospheric  $\text{CH}_4$  markedly varies with the nature and the quantity of fertilisers applied. Ammonium and urea usually inhibit atmospheric  $\text{CH}_4$  oxidation and nitrate does not. In particular, the inhibitory effect of ammonium is well demonstrated [25, 40, 59, 105, 118, 134, 146, 199]. Long-term experiments at Rothamsted in a 1-km<sup>2</sup> area have classified the  $\text{CH}_4$  oxidising potential of soils in the following order: forest > pastures > cultivated soils [241], which indirectly demonstrates a relationship between the quantity of fertiliser applied and the level of  $\text{CH}_4$  oxidation in soils. Similarly, the comparison of thirteen soils of same origin, in Scotland, Denmark and Poland, either planted with trees or cultivated, showed that putting a soil under culture reduced its  $\text{CH}_4$  oxidation activity by about 60 % [156].

In cultivated soils of the Rothamsted Experimental Station, mineral N-fertiliser ( $(\text{NH}_4)_2\text{SO}_4$  and  $\text{KNO}_3$ ), inhibited atmospheric  $\text{CH}_4$  oxidation whereas organic fertilisation did not [85]. Mineral N applied annually as  $(\text{NH}_4)_2\text{SO}_4$ , at 96 or 144 kg N·ha<sup>-1</sup> for 130 years, completely inhibited  $\text{CH}_4$  oxidation, even where lime was applied to maintain a soil pH of about 6. By contrast, the long-term application of N as  $\text{NaNO}_3$  (96 kg N·ha<sup>-1</sup>) caused no decline in  $\text{CH}_4$  oxidation as compared to unfertilised grassland at the same pH; in some cases, it caused a small increase. Withholding  $\text{NH}_4\text{-N}$  for 3 years caused no significant recovery of  $\text{CH}_4$  oxidation; withholding  $\text{NO}_3\text{-N}$  caused a slight decline [86]. In the pasture, the  $\text{CH}_4$  oxidation potential significantly decreased in plots fertilised for 138 years with ammonium fertiliser whereas it was not affected by nitrate fertiliser [241]. The inhibitory effect of N-fertilisation was also observed in soils under dryland rice [195] and in peat soils that were drained and fertilised, inhibition being faster with  $\text{NH}_4\text{Cl}$  as compared to  $\text{KNO}_3$  and urea [43]. The inhibitory of mineral N observed in upland agricultural soils was

also reported in landfill soils (sixty-four inhibition by  $\text{NH}_4\text{NO}_3$ ) [105].

In fertilised forest soils, ammonium had a partial inhibitory effect on atmospheric  $\text{CH}_4$  oxidation. The level of inhibition (15–40 %) was positively correlated either with the quantity of fertiliser applied or the content of the soil in available N [198]. An inverse relationship between N-availability and  $\text{CH}_4$  uptake was also observed in temperate forest soils [202]. In forest soils of Massachusetts, applying 50 and 150 kg  $\text{NH}_4\text{NO}_3\text{-N}\cdot\text{ha}^{-1}\cdot\text{year}^{-1}$  reduced atmospheric  $\text{CH}_4$  oxidation by 15 and 64 % respectively [31]. Urea application in pine forest soils in Florida reduced it by five to twenty times [30].

The inhibition of  $\text{CH}_4$  oxidation in soils by  $\text{NH}_4$  is attributed to a competition at the level of the methane-mono-oxygenase, a transfer of the  $\text{CH}_4$  oxidising activity towards nitrification [30, 146] and the toxicity of  $\text{NO}_2$  produced. Observations with forest soils showed that nitrite, the end product of methanotrophic ammonia oxidation, was a more effective inhibitor of  $\text{CH}_4$  consumption than ammonium [182]. This inhibition only affected atmospheric  $\text{CH}_4$  oxidation and could persist after nitrification of the added  $\text{NH}_4^+$  [146] and become irreversible [109]. The inhibition can be released at  $\text{CH}_4$  concentrations higher than 100 ppm [109].

On the other hand, N-fertiliser application in infertile environments such as acidic meadows, where atmospheric  $\text{CH}_4$  oxidation is negligible, can significantly increase this activity. Methane oxidation activity in a *Calluna* meadow increased from 0.01 to 0.28 mg  $\text{CH}_4\cdot\text{m}^{-2}\cdot\text{d}^{-1}$  after applying 112 kg N·ha<sup>-1</sup>·year<sup>-1</sup> for 6 years, which allowed a significant growth of grasses in the ecosystem [116].

Upland soils are generally behaving as  $\text{CH}_4$  sinks, however some can also behave as  $\text{CH}_4$  sources as observed in a sugarcane soil where chemical fertilisers had effects similar to those observed in wetlands:  $\text{CH}_4$  emission (297 to 1 005 g  $\text{CH}_4\text{-C}\cdot\text{ha}^{-1}$ ) occurred from plots fertilised with urea whereas  $\text{CH}_4$  consumption (442 to 467 g  $\text{CH}_4\text{-C}\cdot\text{ha}^{-1}$ ) was measured in plots fertilised with ammonium sulphate only [233].

## 7.2. Other cultural practices

In upland soils, soil compaction by agricultural equipment may reduce  $\text{CH}_4$  oxidation by half [74]. In cultivated soils in Germany, direct seeding with no ploughing of the soil increased  $\text{CH}_4$  oxidation by six to eight times as compared with ploughed soil [84]. A possible cause of the reduction of  $\text{CH}_4$  oxidising activity in ploughed soils is the destruction of micro-aerophilic niches and the organic matter enriched layer that develops at the top of uncultivated soils [86, 199]. The gas exchange response of different soil types to tillage, particularly  $\text{CH}_4$  oxidation rate, which is affected by long-term soil structural damage, is a potentially useful aspect of soil quality when taken in conjunction with other qualities [8].

## 8. CONCLUSION

### 8.1. Potential ways for mitigation

#### 8.1.1. Cultivated methanogenic soils: ricefields

Cultivated wetlands are mostly ricefields. Strategies to reduce CH<sub>4</sub> emission by ricefields may be oriented toward (i) reducing CH<sub>4</sub> production, (ii) increasing CH<sub>4</sub> oxidation, and (iii) reducing CH<sub>4</sub> transport through the plant. Potential techniques include water, fertiliser management, cropping pattern, varietal selection, and, possibly, the use of selective inhibitors.

##### 8.1.1.1. Water management

Introducing drainage periods during the crop cycle appears to be the most efficient management practice to reduce CH<sub>4</sub> emission from ricefields. Irrigated rice is susceptible to water deficiency during the flowering and grain formation stages [47]; therefore, long drainage period should be avoided during these stages. However, when intermittent drainage is properly used, it does not reduce grain yield [179]. A drainage of a few days at the beginning of the crop cycle favours the anchorage of the young seedling, their growth during tillering and soil N-mineralisation. Drainage also reduces the accumulation of toxic organic acids in the soil and help control vectors of human diseases [170]. It was extrapolated that introducing intermittent drainage periods in 33 % of the poorly drained ricefields in China could reduce by 10 % the agricultural CH<sub>4</sub> emissions (9.9 ± 3.0 Tg) in this country [101]. However, intermittent drainage has also some disadvantages. It may consume two to three times more water than continuous flooding [179]. It may also increase nitrification and N-losses by denitrification and the emission of N<sub>2</sub>O, another greenhouse gas, during resubmersion of the soil [26, 29, 166]. Finally, intermittent drainage requires a good soil levelling and water management facilities that are available only in a small percentage of wetland ricefields [101, 246].

##### 8.1.1.2. Fertilisation and nitrification inhibitors

The methods of fertilisation known to reduce CH<sub>4</sub> emission include: (i) combining organic fertilisers with mineral N-fertiliser; (ii) preferential utilisation of sulphate-containing fertilisers in environments not prone to toxicities due to sulphate reduction; and (iii) deep placement or incorporation of N-fertiliser, which has also additional advantages such as decreasing N-loss by volatilisation, favouring photodependent biological N<sub>2</sub>-fixation, and decreasing the incidence of the vectors of human diseases [170]. Acetylene, brought as encapsulated calcium carbide, increased rice yield by 30 % through its inhibitory effect on nitrification [9] but also decreased CH<sub>4</sub> emission by 35 %. Calcium carbide has therefore a significant potential for adoption.

##### 8.1.1.3. Varietal selection

Rice varietal differences in CH<sub>4</sub> emission of almost 500 % have been reported. Shao and Li [190] defined suitable traits for reducing CH<sub>4</sub> emission as a low level exudation, a large root biomass growing preferentially in the oxidised soil layer, tillers with a structure reducing CH<sub>4</sub> transportation. However such characteristics are the opposite of those required to favour biological N<sub>2</sub> fixation and a high ability to utilise soil N.

##### 8.1.1.4. Crop successions

The cumulative CH<sub>4</sub> emission from tropical rice ecosystems can indeed be lowered by growing suitable upland crops to reduce the submersion period during the annual cropping cycle. Field experiments in India have shown cumulative CH<sub>4</sub> flux of 12–13 g CH<sub>4</sub>·m<sup>-2</sup> from an upland crop followed by a lowland rice crop and 40 g CH<sub>4</sub>·m<sup>-2</sup> in a rice-rice rotation. The seasonal CH<sub>4</sub> emission from the lowland rice grown in the wet season was lower after an upland dry season crop than after a dry season flooded rice [2]. Ratoon rice, which already possesses a developed root biomass, emitted much more CH<sub>4</sub> (540–830 kg CH<sub>4</sub>·ha<sup>-1</sup>) than the initial crop (185–300 kg CH<sub>4</sub>·ha<sup>-1</sup>) [128].

##### 8.1.1.5. Techniques suitable for adoption in rice cultivation

According to the International Rice Research Institute [89], high emission rates are associated with specific management practices, some of which can be modified to reduce emissions without affecting yield. Mitigation strategies that may improve rice productivity can be derived from the following findings [89]:

- Temporary soil aeration reduces CH<sub>4</sub> emission while maintaining rice yields. Temporary soil aeration can also reduce water demand in fields that have an impermeable subsoil. A number of water management strategies have been tested to produce more rice with less water [72]. They rely on reduced water depth and time of submersion. They may also reduce CH<sub>4</sub> emission.
- Modern rice plants are characterised by low root exudation leading to relatively low CH<sub>4</sub> emission rates. Plants grown under sufficient nutrient supply have less root exudation than those grown under nutrient deficiencies, e.g. for phosphorus.
- The increment in CH<sub>4</sub> emission rates triggered by organic manure can be greatly reduced by applying compost residues, which in turn improve soil fertility.
- Methane production and denitrification, the main mechanism for N-losses, are inhibited by a largely identical set of factors. The use of nitrification inhibitors reduces both N-losses and CH<sub>4</sub> emission.
- Direct seeding, instead of transplanting, reduces CH<sub>4</sub> emission with no negative impact on yield.

### 8.1.2. Cultivated upland soils and forests

Upland cultivated soils are CH<sub>4</sub> sinks whose efficiency is markedly reduced by cultural practices.

Methods identified as non or less detrimental for the CH<sub>4</sub> oxidising potential of such soils are (i) using organic [85] and/or nitrate N-fertilisers [241] and (ii) direct seeding with no ploughing [84]. Forest soils are usually efficient CH<sub>4</sub> sinks. When fertilised, they are often treated with urea in the form of super-granules or briquettes of a few grams. Urea application is known to decrease methanotrophy by five to twenty times [30]. This inhibitory effect was avoided when fertilisation was brought as (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> in solution [236].

Ammonium fertilisation of cultivated fields, grasslands and forests causes an irreversible inhibition of the atmospheric CH<sub>4</sub> oxidation potential of these soils. The use of nitrate fertiliser and/or organic fertilisers should be encouraged in such soils.

Obviously, techniques that may contribute to reduced atmospheric CH<sub>4</sub> concentration through the management of cultivated wetland and upland soils must, to be adopted, have a significant advantage for farmers. This reduces the applicability of identified potential methods. Further studies to verify the mitigation options should focus on feasibility for local farmers [146].

### 8.1.3. Non-cultivated soils

Uncultivated soils are in most cases 'orphan sites' with regard to the mitigation of CH<sub>4</sub> emission. Management practices that reduce emission in wetlands and oxidation in uplands will be financed only if they have a significant economical impact in the short term. For example the drainage of malaria-prone swamps or that of peat areas to allow cultivation will contribute to reduce CH<sub>4</sub> emissions. Improving the fertility of acid meadows to allow sheep raising will increase soil methanotrophic activity through the growth of grass [116].

## 8.2. Gaps in knowledge

The main gap in basic knowledge deals with the microflora involved, which is still very imperfectly understood. Recent studies demonstrate the implication of numerous uncultivable strains in the CH<sub>4</sub> cycle. In particular CH<sub>4</sub>-oxidising bacteria responsible for the consumption of atmospheric CH<sub>4</sub> are largely unknown. Micro-organisms that have been isolated and most studied are not necessarily those that are the only or the most active in soil.

The development of methods to establish global and regional budgets of greenhouse gases has been for a significant part driven by international agreements requiring governments to establish emission inventories and to develop means to stabilise or reduce national emissions. Although techniques and models for quantifying gas fluxes have improved considerably for some gases and sources, large uncertainties remain at the national, regional and global budgets [176]. Estimating CH<sub>4</sub> emission in a region or a broad type of environment (i.e. ricefields) requires the measurement of emission rates under a representative set of environmental conditions. However, identifying the factors

that control emissions rates is difficult, and there are uncertainties in determining how many different environmental combinations have to be studied to characterise the source. Specific local emission rates must be extrapolated to regional or global scales, and while scale-specific data may be available, they are much more uncertain than the measured emission rates [104]. As pointed out by Milich [138], "the greatest uncertainty arises in associating measured emission rate with an uncertain extrapolant, even though both the extrapolant and the emission rate may be accurately known. Usually the extrapolant came from data that were obtained for purposes other than global change".

Cultivated wetland areas (ricefields) are generally better characterised and quantified in terms of type of CH<sub>4</sub> source than non-cultivated wetlands but uncertainties still remain. For example, CH<sub>4</sub> emission estimates from Chinese ricefields obtained using several methods suggested emission value of  $13.0 \pm 3.3 \text{ Tg}\cdot\text{year}^{-1}$  and a range from 9.7 to 16.2 Tg·year<sup>-1</sup> [176].

Uncultivated wetlands are often poorly characterised as CH<sub>4</sub> sources and the corresponding area is often difficult to estimate because of:

- the lack of regional data; the work of Barnaud [11] is an example of the difficulties encountered in obtaining geographical quantitative data on natural wetlands in France;
- their large variability in time; for example, in the tropics, variations in precipitation are the major source of seasonal changes and affect sometimes quite dramatically the extent of land inundation [138], which renders extrapolations difficult;
- problems in the legal definitions of wetlands; the national Council for Science and the Environment (USA) has established a web site devoted to 'Wetland Issues' (URL: <http://www.cnie.org/nle/crswet.html>). Among the twelve reports presented, none refers to CH<sub>4</sub>, and the wetland classification used is based mostly on aspects dealing with the conservation of the aquatic environment and wildlife, an approach that is of little use for extrapolating CH<sub>4</sub> emissions.

Despite the uncertainties on the contributions of soils of various cultivated and non-cultivated environments to the CH<sub>4</sub> global budget, it is clear that soils constitute at the global scale a major CH<sub>4</sub> source: production in wetlands is obviously much larger than consumption in uplands. On a short-term basis, the most promising approach to decreasing production by soils is obviously through adequate water management of ricefields. A large range of other approaches has been identified, their potential for adoption will primarily depend on economical aspects.

## 9. RELATED WEB SITES

Additional information can be found at the following WEB sites:

The Intergovernmental Panel on Climate Change (IPCC): <http://www.ipcc-nggip.iges.or.jp/>  
 Publications IPCC: <http://www.ipcc.ch/pub/pub.htm>  
 The World Resource Institute: <http://www.wri.org/>  
 National Council for Science and the Environment: <http://www.cnie.org/>  
 National Library for the Environment: <http://www.cnie.org/nle/>  
 United States Environmental Protection Agency: <http://www.epa.gov/globalwarming/>  
 MIT Joint Program on the Science and Policy of Global Change: <http://salticus-peckhamae.mit.edu/afs/athena.mit.edu/org/g/globalchange/www/>  
 International Rice Research Institute (IRRI): <http://www.cgiar.org/irri/>

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